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10/820,822	04/09/2004	Hae-Kyoung Kim	61610134US	8493

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VIENNA, VA 22182

EXAMINER

WANG, EUGENIA

ART UNIT	PAPER NUMBER
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1795

NOTIFICATION DATE	DELIVERY MODE
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07/16/2008

ELECTRONIC

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

PATENT@PARK-LAW.COM

DETAILED ACTION

Response to Amendment

1. In response to the amendment received May 14, 2008:
 - a. Claim 21 has been cancelled as per Applicant's request. Claims 1, 3, 4, 6, 7, 9-20, 22, and 23 are pending.
 - b. The core of the previous rejection has been maintained with any changes necessitated by the amendment.

Continued Examination Under 37 CFR 1.114

2. A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on May 14, 2008 has been entered.

Claim Objections

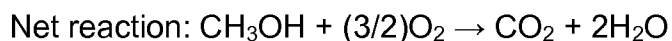
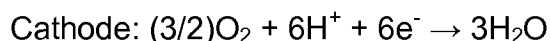
3. Claims 10, 12, 14, 16, and 23 are objected to because of the following informalities: claim 10 recites that "the diluent is H₂O," wherein examiner submits that 'comprises' is a more appropriate term than "is."

It is noted that, as recited in the office action dated October 29, 2007, that this language is given the interpretation that the diluent comprises H₂O. This is because there is no support within the Specification that the diluent provided consists of water.

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As stated in paragraph 0022: "The byproduct of the oxidation and reduction reactions, for example, water, flows into the diluent storage unit 107 and is used as fuel diluent."

It is noted that the reaction for a direct methanol fuel cell are as follows:



So, therefore, at least CO_2 would be a byproduct stored in the diluent tank as well and could be considered as diluent. Furthermore, the reactions listed above are *theoretical* reactions, considering 100% consumption of methanol. No support exists to show that 100% consumption positively exists. In this manner, one of ordinary skill would reasonably expect some unconsumed fuel and oxidant to be released in the exhaust. Furthermore, no separator is shown between the exhaust and the diluent storage tank for separating unreacted reactants and the water (looking at fig. 1). Therefore, the broadest reasonable interpretation of the new limitation is comprising language, barring proof to the contrary.

Appropriate correction is required.

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States

only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

4. Claim 17 is rejected under 35 U.S.C. 102(e) as being anticipated by US 6890674 (Beckmann et al.).

As to claim 17, Beckmann et al. teach the use of Nafion, which expands (changes volume) relative to methanol concentration is used as a switch, valve, or sensor in a fuel cell (col. 8, lines 7-25). It is shown in fig. 7A and 7B that the sensor film is on a substrate. Fig. 8 depicts the sensor embodiment having a conductor [70] fastened to Nafion material [72]. The sensor embodiment of Nafion communicates a concentration level of methanol (thus outputting a signal) (col. 8, lines 37-38). The Nafion conductor displays such a signal via known resistance values, wherein relaxed and strained Nafion have different resistance values. Beckmann et al. embodies an electrical signal, as it is listed that the sensor embodied in fig. 8 has a current running through the conductor and that a Wheatstone bridge circuit may be used to determine such a resistance (col. 8, lines 37-51 and col. 8, 61-67). Since the methanol concentration affects the relax and strain in Nafion, it thus sends the resistance values in comparison to known values. In this manner, a signal is output with respect to an expansion coefficient. These signals sent correspond to values that can be interpreted as both within and not within a defined reference range (barring clear definition of what constitutes a reference range). Furthermore, Beckmann et al. teach that Nafion expansion and resistance are both proportional to methanol concentration (col. 8, lines 57-60). Accordingly, a reference range with respect to resistance inherently corresponds to a reference range of expansion coefficients of the sensor film.

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Additionally, the sensor as taught by Beckmann et al. would be capable of sending out a signal when an expansion coefficient of the sensor is not within a reference range of expansion coefficients of the sensor film, as it outputs signals with respect to the expansion of Nafion (see the second embodiment in col. 8).

It has been held that the recitation of an element is “capable” of performing a function is not a positive limitation but only requires the ability to so perform. It does not constitute a limitation in any patentable sense. *In re Hutchinson*, 69 USPQ 138.

While intended use recitations and other types of functional language cannot be entirely disregarded. However, in apparatus, article, and composition claims, intended use must result in a structural difference between the claimed invention and the prior art in order to patentably distinguish the claimed invention from the prior art. If the prior art structure is capable of performing the intended use, then it meets the claim. In a claim drawn to a process of making, the intended use must result in a manipulative difference as compared to the prior art. *In re Casey*, 370 F.2d 576, 152 USPQ 235 (CCPA 1967); *In re Otto*, 312 F.2d 937, 938, 136 USPQ 458, 459 (CCPA 1963).

Claims directed to apparatus must be distinguished from the prior art in terms of structure rather than function. *In re Danly*, 263 F.2d 844, 847, 120 USPQ 528, 531 (CCPA 1959). See also MPEP § 2114.

The manner of operating the device does not differentiate an apparatus claim from the prior art. A claim containing a “recitation with respect to the manner in which a claimed apparatus is intended to be employed does not differentiate the claimed apparatus from a prior art apparatus” if the prior art apparatus teaches all the structural

limitations of the claim. Ex parte Masham, 2 USPQ2d 1647 (Bd. Pat. App. & Inter. 1987).

Lastly, it is noted that, since the electrical signal (the resistance, which corresponds to the use of a conductor with current running through it, which can be monitored by a circuit) of Beckmann et al. is based on (consists of) a Nafion sensor which expands (increases in volume) with respect to the fuel concentration (see col. 8, lines 8-25), the signal in Beckman et al.'s system is inherently based on a variable input consisting of the concentration of the fuel and the volume of the sensor film, since the volume of the sensor film is related to the concentration of methanol.

Where applicant claims a composition in terms of a function, property or characteristic and the composition of the prior art is the same as that of the claim but the function is not explicitly disclosed by the reference, the examiner may make a rejection under both 35 U.S.C. 102 and 103, expressed as a 102/103 rejection.

The fact that a certain result or characteristic may occur or be present in the prior art is not sufficient to establish the inherency of that result or characteristic. In re Rijckaert, 9 F.3d 1531, 1534, 28 USPQ2d 1955, 1957 (Fed. Cir. 1993).

“In relying upon the theory of inherency, the examiner must provide a basis in fact and/or technical reasoning to reasonably support the determination that the allegedly inherent characteristic necessarily flows from the teachings of the applied prior art.” Ex parte Levy, 17 USPQ2d 1461, 1464 (Bd. Pat. App. & Inter. 1990)

In the case of the instant application the basis for expectation of inherency is that since the electrical signal of resistance (as taught by Beckmann et al.) is dependent on

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the volume of the Nafion, which in turn is dependent the concentration of the methanol fuel (col. 8, lines 8-25 and 37-60). Therefore, since the signal is dependent on the volume, which is related to the concentration, the signal of Beckmann et al. is based on (consists of) the input parameters of both volume of the sensor film and concentration of the film.

The Examiner requires applicant to provide that the prior art products do not necessarily or inherently possess the characteristics of his [or her] claimed product.

Whether the rejection is based on inherency' under 35 U.S.C. 102, on prima facie obviousness' under 35 U.S.C. 103, jointly or alternatively, the burden of proof is the same...[footnote omitted]." The burden of proof is similar to that required with respect to product-by-process claims. In re Fitzgerald, 619 F.2d 67, 70, 205 USPQ 594, 596 (CCPA 1980) (quoting In re Best, 562 F.2d 1252, 1255, 195 USPQ 430, 433-34 (CCPA 1977)).

5. Claim 18 is rejected under 35 U.S.C. 102(e) as being anticipated by Beckmann et al. as evidenced by US 2003/0091887 (Ihonen et al.) and DuPont – Nafion Membranes and Dispersions.

As to claim 18, Nafion (the material taught by Beckmann et al.) is proton conducting polymer (as evidenced by Ihonen et al. (para 0003) and Nafion Membranes and Dispersions).

Response to Arguments

6. Applicant's arguments filed May 14, 2008 have been fully considered but they are not persuasive.

Applicant argues that Beckmann et al. does not teach an electrical signal that is determined based on a variable input consisting of the concentration of fuel and the volume of the fuel sensor, since Beckmann et al. teaches of using volume change of Nafion to vary the resistance of the conductor in order to sense the methanol concentration.

Examiner respectfully disagrees and submits that such a variable input as claimed is inherent, as explained within the rejection and reiterated herein for clarity's sake. First of all, since Beckmann et al. embodies a sensor as in fig. 8, has a current running through the conductor, and teaches that a Wheatstone bridge circuit may be used to determine such a resistance (col. 8, lines 37-51 and col. 8, 61-67), the signal is viewed as being electrical. Examiner notes that Applicant admits that the sensor changes volume (bottom of page 8 to top of page 9 in arguments). However, since the volume change is inherently dependent on methanol concentration, the variable input consists of the concentration of fuel and volume of the sensor. In the case of the instant application the basis for expectation of inherency is that since the electrical signal of resistance (as taught by Beckmann et al.) is dependent on the volume of the Nafion, which in turn is dependent on the concentration of the methanol fuel (col. 8, lines 8-25 and 37-60). Therefore, since the signal is dependent on the volume, which is related to the concentration, the signal of Beckmann et al. is based on (consists of) the input parameters of both volume of the sensor film and concentration of the film. Since Applicant fails to show proof how this interpretation fails to teach the claimed invention or is incorrect, the rejection is upheld.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

The factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

1. Determining the scope and contents of the prior art.
2. Ascertaining the differences between the prior art and the claims at issue.
3. Resolving the level of ordinary skill in the pertinent art.
4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

7. Claims 1, 3, 4, 6, 7, 9, 11, 13, 15, and 22 are rejected under 35 U.S.C. 103(a) as being unpatentable over US 6303244 (Surampudi et al.) in view of Beckmann et al.

As to claim 1, Surampudi et al. discloses a direct methanol feed fuel cell system. The system is composed of a fuel cell stack [924], a methanol fuel storage tank [900], a circulating tank [906], condensers [940, 942] (which acts as a diluent storage unit that stores only a diluent that is a byproduct of the chemical reaction in the fuel cell stack, wherein the diluent comprises water (col. 18, lines 31-39), and a methanol concentration sensor that provides input to a controller to regulate the fuel cell system (col. 18 lines 5-19; See Figure 9). The fuel cell stack is comprised of an anode and cathode and generates electrical energy (col. 3 lines 25-32).

Surampudi et al. does not disclose that the sensor comprises a sensor film or sensor member that changes volume thereof depending on the concentration of the fuel, wherein the signal is determined based on a variable input consisting of the concentration of the fuel and volume of the sensor film.

Beckmann et al. teaches a method and apparatus for managing fluids in a fuel cell system. Beckmann teaches the use of various devices to control fuel concentration in a direct oxidation fuel cell system such as a direct methanol fuel cell (col. 1 lines 39-42; col. 2 line 63 to col. 3 line 4). One device for determining the concentration of the fuel is a sensor (col. 3 lines 50-62). The sensor is constructed of Nafion™ (serves as sensor film and sensor member), wherein Nafion™ expands or varies in volume when exposed to a methanol solution (col. 8 lines 8-16). The amount of expansion experienced by the Nafion™ is directly related to the concentration of methanol fuel. The amount Nafion™ expands is predictable and essentially linear over the relevant methanol concentrations (col. 8 lines 21-25). Therefore, the sensor embodiment of

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Nafion communicates a concentration level of methanol (thus outputting a signal) (col. 8, lines 37-38). The Nafion conductor displays such a signal via known resistance values, wherein relaxed and strained Nafion have different resistance values. Lastly, it is noted that, since the signal (which corresponds to the resistance) of Beckmann et al. is based on (consists of) a Nafion sensor which expands (increases in volume) with respect to the fuel concentration (see col. 8, lines 8-25), the signal in Beckman et al.'s system is inherently based on a variable input consisting of the concentration of the fuel and the volume of the sensor film, since the volume of the sensor film and concentration of methanol are related.

Where applicant claims a composition in terms of a function, property or characteristic and the composition of the prior art is the same as that of the claim but the function is not explicitly disclosed by the reference, the examiner may make a rejection under both 35 U.S.C. 102 and 103, expressed as a 102/103 rejection.

The fact that a certain result or characteristic may occur or be present in the prior art is not sufficient to establish the inherency of that result or characteristic. In re Rijckaert, 9 F.3d 1531, 1534, 28 USPQ2d 1955, 1957 (Fed. Cir. 1993).

"In relying upon the theory of inherency, the examiner must provide a basis in fact and/or technical reasoning to reasonably support the determination that the allegedly inherent characteristic necessarily flows from the teachings of the applied prior art." Ex parte Levy, 17 USPQ2d 1461, 1464 (Bd. Pat. App. & Inter. 1990)

In the case of the instant application the basis for expectation of inherency is that since the electrical signal of resistance (as taught by Beckmann et al.) is dependent on

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the volume of the Nafion, which indicates the concentration of the methanol fuel (col. 8, lines 8-25 and 37-60). Therefore, since the signal is dependent on the volume, which is related to the concentration, the signal of Beckmann et al. is based on (consists of) the input parameters of both volume of the sensor film and concentration of the film.

The Examiner requires applicant to provide that the prior art products do not necessarily or inherently possess the characteristics of his [or her] claimed product.

Whether the rejection is based on inherency' under 35 U.S.C. 102, on prima facie obviousness' under 35 U.S.C. 103, jointly or alternatively, the burden of proof is the same...[footnote omitted]." The burden of proof is similar to that required with respect to product-by-process claims. In re Fitzgerald, 619 F.2d 67, 70, 205 USPQ 594, 596 (CCPA 1980) (quoting In re Best, 562 F.2d 1252, 1255, 195 USPQ 430, 433-34 (CCPA 1977)).

The motivation to use the concentration sensor is to accurately measure and control the methanol concentration provided to the fuel cell. Furthermore, one of ordinary skill in the art would have been able to appreciate the use of the concentration sensor as taught by Beckmann et al. in the system of Surampudi et al. with reasonable expectation of success. Therefore it would have been obvious to one having ordinary skill in the art at the time the claimed invention was made to modify the system of Surampudi et al. to include Nafion™ (a material that varies in volume depending on the concentration of the methanol solution to which it is exposed) as taught by Beckmann et al. in order to accurately measure and control the methanol concentration provided to

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the fuel cell, as one of ordinary skill in the art would have appreciated the changing of methanol sensors with reasonable expectation of success.

As to claim 3, Surampudi et al. teach of a fuel mixing unit (circulation tank [906]). Circulation tank [906] is a fuel mixing unit, as it allows the diluent from condenser [940, 942] to flow into it as well as methanol from the fuel storage unit (methanol tank [900]).

As to claim 4, Surampudi et al. teaches that the methanol sensor should be located in the methanol fuel or very close to the methanol fuel (col. 18 lines 14-15).

As to claim 6, Surampudi et al. teach a line between the fuel storage unit (methanol tank [900]) and the diluent storage unit (condensers [940, 942]). This line is [918], and it supplies the fuel mixture to the fuel cell stack (fig. 9).

As to claim 7, Surampudi et al. show sensor [916] is located in line [918] (fig. 9).

As to claim 9, the combination of Surampudi et al. with Beckmann et al. teaches that the sensor comprises a substrate and a sensor film attached to a surface of the substrate, since Beckmann et al. teach the sensor film is on a substrate, as shown in fig. 7A and 7B.

As to claim 13, Beckmann et al. teaches a sensor using Nafion embodied in fig. 8. In this embodiment, the sensor is an electronic circuit that outputs signals depending on the change in the volume sensor. This is done as Nafion communicates a concentration level of methanol (col. 8, lines 37-38). The Nafion conductor displays such a signal via known resistance values, wherein relaxed and strained Nafion have different resistance values. These signals sent are in some way electronic, as the sensor is an electronic circuit (fig. 8; col. 8, lines 37-67).

As to claim 22, Beckmann et al.'s yields the control system claimed. Fig. 8 depicts the sensor embodiment having a conductor [70] fastened to Nafion material [72] (sensor film). The sensor embodiment of Nafion communicates a concentration level of methanol (thus outputting a signal) (col. 8, lines 37-38). The Nafion conductor displays such a signal via known resistance values, wherein relaxed and strained Nafion have different resistance values. The methanol concentration affects this and thus sends the resistance values in comparison to known values. In this manner, a signal is output with respect to an expansion coefficient. These signals sent correspond to values that can be interpreted as both within and not within a defined reference range (barring clear definition of what constitutes a reference range). Furthermore, Beckmann et al. teach that Nafion expansion and resistance are both proportional to methanol concentration (col. 8, lines 57-60). Accordingly, a reference range with respect to resistance inherently corresponds to a reference range of expansion coefficients of the sensor film. Additionally, the sensor as taught by Beckmann et al. would be capable of sending out a signal when an expansion coefficient of the sensor is not within a reference range of expansion coefficients of the sensor film, as it outputs signals with respect to the expansion of Nafion (see the second embodiment in col. 8).

It has been held that the recitation of an element is "capable" of performing a function is not a positive limitation but only requires the ability to so perform. It does not constitute a limitation in any patentable sense. *In re Hutchinson*, 69 USPQ 138.

While intended use recitations and other types of functional language cannot be entirely disregarded. However, in apparatus, article, and composition claims, intended

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use must result in a structural difference between the claimed invention and the prior art in order to patentably distinguish the claimed invention from the prior art. If the prior art structure is capable of performing the intended use, then it meets the claim. In a claim drawn to a process of making, the intended use must result in a manipulative difference as compared to the prior art. In re Casey, 370 F.2d 576, 152 USPQ 235 (CCPA 1967); In re Otto, 312 F.2d 937, 938, 136 USPQ 458, 459 (CCPA 1963).

Claims directed to apparatus must be distinguished from the prior art in terms of structure rather than function. In re Danly, 263 F.2d 844, 847, 120 USPQ 528, 531 (CCPA 1959). See also MPEP § 2114.

The manner of operating the device does not differentiate an apparatus claim from the prior art. A claim containing a “recitation with respect to the manner in which a claimed apparatus is intended to be employed does not differentiate the claimed apparatus from a prior art apparatus” if the prior art apparatus teaches all the structural limitations of the claim. Ex parte Masham, 2 USPQ2d 1647 (Bd. Pat. App. & Inter. 1987).

8. Claims 11 and 15 are rejected under 35 U.S.C. 103(a) as being unpatentable over Surampudi et al. in view of Beckmann et al., as applied to claims 1 and 9, as evidenced by Ihonen et al. and DuPont – Nafion Membranes and Dispersions.

As to claims 11 and 15, Nafion (the material taught by Beckmann et al.) is proton conducting polymer that is a perfluorinated sulfonic acid polymer (as evidenced by Ihonen et al. (para 0003) and Nafion Membranes and Dispersions).

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9. Claim 19 rejected under 35 U.S.C. 103(a) as being unpatentable over Beckmann et al., as applied to claim 17, in view of Surampudi et al.

With respect to claim 19, Beckmann et al. teaches the use of Nafion as the sensor but does not teach the use of polystyrene sulfonic acid, poly ether ether sulfone sulfonic acid, sulfonated polyolefin, or sulfonated polysulfone as the polymeric ion exchange membrane in the sensor.

Surampudi et al. demonstrates that Nafion and polyethylene and polypropylene sulfonic acids (sulfonated polyolefins) and polystyrene sulfonic acids are function equivalents within the use of a fuel cell (col. 6, lines 55-57). Therefore, one of ordinary skill in the art at the time the invention was made would have found it obvious to substitute the Nafion of the sensor of Beckmann et al. with sulfonated polyolefins or polystyrene sulfonic acids, as taught by Surampudi et al., with predictable result of obtaining a sensor that functioned in the same manner. It has been held to be within the general skill of a worker in the art to select a known material on the basis of its suitability for the intended use as a matter of obvious design choice. *In re Leshin*, 125 USPQ 416.

Response to Arguments

10. Applicant's arguments filed May 14, 2008 have been fully considered but they are not persuasive.

Applicant argues that the combination does not teach (1) a sensor that detects a concentration of a fuel in a fuel mixture solution and outputs a signal according to the concentration, (2) a control unit that receives the signal from the sensor and controls the

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fuel mixture solution, wherein (3) the diluent comprises H₂O and wherein the sensor comprises a sensor film that changes volume thereof depending on the concentration of the fuel, wherein (4) the signal is determined based on a variable input consisting of the concentration of the fuel and volume of the sensor.

Examiner respectfully disagrees. First of all, Applicant makes the conclusory statement that the prior art fails to teach the listed features of claim 1. However, there is no proof or reasoning as to how the combination fails to teach such limitations.

Additionally, Examiner submits that the combination of Surampudi et al. and Beckmann et al. teaches the limitations, as set forth above in the 103 rejection (reiterated below for clarity's sake).

With respect to (1) and (2), Examiner has set forth that Surampudi et al. disclose a direct methanol feed fuel cell system wherein there is a methanol concentration sensor that provides input to a controller to regulate the fuel cell system (col. 18 lines 5-19; See Figure 9). (The input to the controller is seen as the signal made by the concentration sensor.) Therefore, Surampudi et al. teach of the sensor that outputs a signal according to the concentration, wherein such a control unit receives the signal.

With respect to (3), Examiner notes that Surampudi et al. disclose that condensers [940, 942] act as a diluent storage unit that stores only a diluent that is a byproduct of the chemical reaction in the fuel cell stack, wherein the diluent comprises water (col. 18, lines 31-39; fig. 9). At this point, Beckmann et al. is relied upon to teach the claimed sensor. The sensor is constructed of Nafion™ (serves as sensor film and sensor member), wherein the Nafion™ expands or varies in volume when exposed to a

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methanol solution (col. 8 lines 8-16). The amount of expansion experienced by the Nafion™ is directly related to the concentration of methanol fuel. The amount Nafion™ expands is predictable and essentially linear over the relevant methanol concentrations (col. 8 lines 21-25). Therefore, Surampudi et al. disclose the claimed diluent, and the combination with Beckmann et al. shows the sensor as claimed. (It is noted that the reasons for combination are set forth within the rejection, wherein such combination is seen as valid, since Applicant has not particularly pointed out why the combination is not valid.)

With respect to (4), Examiner submits that such a variable input as claimed is inherent. First of all, since Beckmann et al. also teaches a concentration sensor wherein resistance (signal) is used to communicate the concentration level (col. 8, lines 38-60). Examiner notes that Applicant admits that the sensor changes volume (bottom of page 8 to top of page 9 in arguments). However, since the volume change is inherently dependent on methanol concentration, the variable input consists of the concentration of fuel and volume of the sensor. In the case of the instant application the basis for expectation of inherency is that since the signal of resistance (as taught by Beckmann et al.) is dependent on the volume of the Nafion, which in turn is dependent on the concentration of the methanol fuel (col. 8, lines 8-25 and 37-60). Therefore, since the signal is dependent on the volume, which is related to the concentration, the signal of Beckmann et al. is based on (consists of) the input parameters of both volume of the sensor film and concentration of the film.

Accordingly, by replacing the sensor of Surampudi et al. with the sensor of Beckmann et al., as set forth in the rejection, Examiner submits that the claimed limitations are met. Again, since no proof or reasoning as to how the combination fails to teach the claimed limitations is offered, the rejection of record is seen as proper and is upheld.

With respect to the arguments regarding the 103 rejections, Applicant argues that the prior art used to obviate the dependent claims (claims 3, 4, 6, 7, 9, 11, 13, 15, 19, and 22) do not cure the deficiencies of the reference(s) used to reject the independent claims (claims 1 and 17). Applicant does not argue how the combinations are not proper. Therefore, the Examiner maintains the obviousness rejections and upholds the rejection, as above.

Allowable Subject Matter

11. Claims 10, 12, 14, 16, and 23 are objected to (as set forth above), but would be allowable if rewritten to correct the claim objection.

In the non-final office action mailed on October 29, 2007, Examiner has already set for the reasons for allowance of claim 10. Since claims 12, 14, 16, and 23 are dependent on claim 10, they are allowable for the same reasons.

It is noted that claim 10 was amended in such a manner that it does not currently include the previous limitations of claim 1 (its previous base claim). It now includes problematic claim language, which was mentioned in the October 29, 2007 office action.

12. Claim 20 is allowed.

In the non-final office action mailed on October 29, 2007, Examiner has already set for the reasons for allowance of claim 20.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to EUGENIA WANG whose telephone number is (571)272-4942. The examiner can normally be reached on 7 - 4:30 Mon. - Thurs., EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached on 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/E. W./
Examiner, Art Unit 1795

/PATRICK RYAN/
Supervisory Patent Examiner, Art Unit 1795

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